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Synthesis of the (α , α -Difluoroalkyl)phosphonate Analogue of Phosphoserine

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Abstract: The synthesis of the $(\alpha, \alpha$ -difluoroalkyl)phosphonate analogue of L-phosphoserine, 5, in a form appropriate for solid phase peptide synthesis, is reported. Two independent routes are described, starting from L-serine or (R)-isopropylideneglycerol. In each case, PCF₂-C bond formation is achieved by triflate displacement with diethyl lithiodifluoromethylphosphonate.

There is currently considerable interest in $(\alpha, \alpha$ -difluoroalkyl)phosphonates as analogues of naturally occurring phosphates.¹ Data in the literature support the notion that phosphate mimics of this class do indeed bind to enzymatic phosphate binding sites. For example, 1 is a potent bisubstrate analogue inhibitor of purine nucleoside phosphorylase.² 2 is a substrate for glycerol 3-phosphate dehydrogenase,³ and 3 is an irreversible inhibitor of EPSP synthase.⁴ More recently, the $(\alpha, \alpha$ -difluoroalkyl)phosphonate analogue of phosphotyrosine 4, was synthesized by two groups^{5,6} and has been incorporated into peptides.^{5a}



The members of the protein phosphoserine/threonine phosphatase class of enzymes, PP1, PP2A, PP2B and PP2C, are important mediators of signal transduction events.⁷ For example, PP2B (calcineurin) is the common target of the immunosuppressants cyclosporin and FK-506.⁸ Enzymes in this family are known to dephosphorylate peptides, in addition to their usual protein substrates.⁹ Therefore, peptides containing an effective, but hydrolytically stable, phosphoserine mimic, are potential inhibitors of this class of enzymes. We report herein the synthesis of 5, a phosphoserine mimic bearing protecting groups amenable to automated solid phase peptide synthesis.^{5a}

Recently, we discovered that $(\alpha, \alpha$ -diffuoroalkyl)phosphonates may be conveniently synthesized via the direct displacement of primary alkyl triflates by diethyl lithiodifluoromethylphosphonate (6).^{10a} We sought to apply this methodology to the synthesis of 5. Initially, our approach was to construct a D- or L-serinol derived triflate and to examine its triflate displacement chemistry. We were pleased to find that triflates **7a-9a** could all be synthesized from the corresponding alcohols and were stable to silica gel chromatography. Furthermore, **7a-9a** do undergo displacement with **6** in 35-70% yields to provide the corresponding (α, α -

difluoroalkyl)phosphonates, **7b-9b**. Unfortunately, in all three cases, attempts to remove the nitrogen protecting group [2,5-dimethylpyrrole (**7b**),¹¹ N-tosyl (**8b**),¹² and N-benzyl (**9b**)¹³] were unsuccessful.



This problem was solved by using a readily removable *tert*-butyldimethylsilyl protecting group¹⁴ for the carbamate nitrogen (see Scheme 1). Thus, starting from the known L-serine derivative 10,¹⁵ esterification, Boc removal and ester reduction (NaBH₄) proceed smoothly. The resulting (R)-O-benzylserinol is transformed into the corresponding oxazolidinone 11 using triphosgene as a convenient phosgene source.¹⁶ N-Silylation¹⁷ and benzyl ether hydrogenolysis yield alcohol 12. The corresponding triflate, being quite labile, is generated, and then immediately subjected to our usual displacement conditions¹⁰ to provide (α,α -difluoroalkyl)phosphonate 13. In this case, N-deprotection is facile (13:1 MeOH/1N HCl rt, 12 h).¹⁴ Installation of the requisite N-Boc protecting group and oxazolidinone ring-opening are accomplished by a modification (EtOH as solvent) of the procedure reported by Ishizuka and Kunieda, in quite modest yield.¹⁸ Finally, four-electron oxidation under the Corey-Schmidt conditions (6 equivalents of PDC)¹⁹ yields the desired, protected phosphoserine analog 5.²⁰



Scheme 1

Parallel to these studies, we explored a complementary route to 5 which begins from the readily available chiral building block $15.^{21}$ Strategically, this approach differs from the previous route, in that here the α -arnino group is installed late in the synthesis, and in this way the N-protecting group issue in circumvented. Triflate synthesis and displacement¹⁰ proceed cleanly to give phosphonate $16.^{22}$ Acetonide cleavage is achieved with ethanol, containing Dowex-50 (rt, 2 days). Selective silvlation of the primary hydroxyl then gives 17. Formation of the secondary triflate is followed by displacement with sodium azide (5 equiv. NaN₃, rt, 2 h).

Azide hydrogenolysis in the presence of Boc_2O^{23} yields the necessary Boc-protected aminophosphonate 19, directly. Chemoselective cleavage of the TBS ether, in the presence of the N-Boc and diethyl difluoromethylphosphonate functional groups, is best achieved using the mildly Lewis acidic conditions described by Lipshutz.²⁴ Oxidation, as before, yields the title compound.



In summary, we have described the synthesis of L-phosphoserine analog 5, by two independent routes. Both emanate from readily available building blocks out of the chiral pool. In each case, the critical PCF₂-C bond is fashioned by the direct displacement of a primary triflate with 6.10 The synthesis of 5 from (R)-isopropylideneglycerol is clearly the more efficient procedure, at this time. Studies on the effectiveness of 5 as a hydrolytically stable phosphoserine mimic will be reported in due course.

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- 20. For 5: ¹H NMR (360 MHz, CDCl₃): δ 1.34-1.38 (t, J = 7 Hz, 6 H), 1.42 (s, 9 H), 2.56-2.75 (m, 2 H), 4.22-4.30 (app quintet, J = 7 Hz, 4 H), 4.55-4.59 (m, 1 H), 5.37-5.40 (m, 1 H); ¹³C NMR (125 MHz, CDCl₃): δ 16.27, 16.30 (2 X OCH₂CH₃); 28.3 (OC<u>Me₃</u>); 35.3-35.4 (br, CH₂CF₂); 48.4 (C_α); 65.0-65.1 (br, 2 X OCH₂CH₃); 80.5 (OCMe₃); 155.5 (Me₃COC₂=0); 173.0 (CO₂H); HRMS (FAB, 3-NOBA/Na₂CO₃) calcd for C₁₃H₂4F₂NO₇PNa (M+Na)⁺ 398.1156, obsd 398.1155; [α]²⁵D -10.8° (c 2.20, MeOH). 5 obtained from (R)-isopropylideneglycerol is ≥ 96% ee as judged by ¹H NMR [A single Mosher ester is seen (500 MHz, CDCl₃) following the sequence: (a) CH₂N₂; (b) HCl, EtOAc; (c) (S)-α-methoxy-α-(trifluoromethyl)phenylacetyl chloride].
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