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## Synthesis of the  $(\alpha, \alpha$ -Difluoroalkyl)phosphonate Analogue of **Phosphoserine**

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Abstract: The synthesis of the  $(\alpha, \alpha$ -difluoroalkyl)phosphonate analogue of L-phosphoserine, 5, in a form appropriate for solid phase peptide synthesis, is reported. Two independent routes **are described, starting from L-serine or (R)-isopropylideneglycerol. In each case, PCFz-C bond formation is achieved by triflate displacement with diethy lithiodifhmromethylphosphonate.** 

There is currently considerable interest in  $(\alpha, \alpha$ -difluoroalkyl)phosphonates as analogues of naturally **occurring phosphates. 1 Data in the literature support the notion that phosphate mimics of this class do indeed bind to enzymatic phosphate binding sites. For example,** 1 **is a potent bisubsuate analogue inhibitor of purine**  nucleoside phosphorylase.<sup>2</sup> 2 is a substrate for glycerol 3-phosphate dehydrogenase,<sup>3</sup> and 3 is an irreversible inhibitor of EPSP synthase.<sup>4</sup> More recently, the  $(\alpha, \alpha$ -difluoroalkyl)phosphonate analogue of phosphotyrosine 4, was synthesized by two groups<sup>5,6</sup> and has been incorporated into peptides.<sup>5a</sup>



**The members of the protein phosphoserine/threonine phosphatase class of enzymes, PPl, PP2A, PPZB and PP2C. are important mediators of signal transduction events. 7 For example, PP2B (calcineurin) is the common target of the immunosuppressants cyclosporin and FK-506.8 Enzymes in this family are known to dephosphorylate peptides, in addition to their usual protein substrates.9 Therefore, peptides containing an effective, but hydrolyticaliy stable, phosphoserine mimic, are potential inhibitors'of this class of enzymes. We**  report herein the synthesis of 5, a phosphoserine mimic bearing protecting groups amenable to automated solid phase peptide synthesis.<sup>5a</sup>

Recently, we discovered that  $(\alpha, \alpha$ -difluoroalkyl)phosphonates may be conveniently synthesized via the direct displacement of primary alkyl triflates by diethyl lithiodifluoromethylphosphonate (6).<sup>10a</sup> We sought to apply this methodology to the synthesis of 5. Initially, our approach was to construct a D- or L-serinol derived **hiflate and to examine its triflate displacement chemistry. We were pleased to find that triflates 7a-9a could all be synthesized from the corresponding alcohols and were stable to silica gel chromatography. Furthermore, 7a-9a** do undergo displacement with 6 in 35-70% yields to provide the corresponding  $(\alpha, \alpha - \alpha)$ 

**difluomalkyl)phosphonates. 7b-9b. Unfortunately, in all three cases, attempts to remove the nitrogen protecting**  group [2,5-dimethylpyrrole  $(7b)$ ,<sup>11</sup> N-tosyl  $(8b)$ ,<sup>12</sup> and N-benzyl  $(9b)$ <sup>13</sup>] were unsuccessful.



**This problem was solved by using a readily removable rert-butyldimethylsilyl protecting group14 for the carbamate nitrogen (see Scheme 1). Thus, starting from the known L-serine derivative** 10.15 **esterification. Boc femoval and ester reduction (NaBH4) proceed smoothly. The resulting (R)-O-benzylserinol is transformed into** the corresponding oxazolidinone 11 using triphosgene as a convenient phosgene source.<sup>16</sup> N-Silylation<sup>17</sup> and **benzyl ether hydrogenolysis yield alcohol 12. The corresponding triflate, being quite labile, is generated, and**  then immediately subjected to our usual displacement conditions<sup>10</sup> to provide  $(\alpha, \alpha$ -difluoroalkyl)phosphonate **13. In this case, N-deprotection is facile (13:l MeOWlN HCl rt, 12 h).14 Installation of the requisite N-Boc**  protecting group and oxazolidinone ring-opening are accomplished by a modification (EtOH as solvent) of the **procedure reported by Ishizuka and Kunieda, in quite modest yield. 18 Finally, four-electron oxidation under the**  Corey-Schmidt conditions (6 equivalents of PDC)<sup>19</sup> yields the desired, protected phosphoserine analog 5.20



**Scheme 1** 

**Parallel to these studies, we explored a complementary route to 5 which begins from the readily available chiral building block 15.21 Strategically. this approach differs from the previous route, in that hem the a-amino group is installed late in the synthesis, and in this way the N-protecting group issue in circumvented. Triflate**  synthesis and displacement<sup>10</sup> proceed cleanly to give phosphonate  $16.22$  Acetonide cleavage is achieved with **ethanol, containing Dowex-50 (rt, 2 days). Selective silylation of the primary hydroxyl then gives 17.**  Formation of the secondary triflate is followed by displacement with sodium azide (5 equiv. NaN<sub>3</sub>, rt, 2 h).

Azide hydrogenolysis in the presence of Boc<sub>2</sub>O<sup>23</sup> yields the necessary Boc-protected aminophosphonate 19, **directly. Chemoselective cleavage of the TBS ether, in the presence of the N-Boc and diethyl difluoromethylphosphonate functional groups, is best achieved using the mildly Lewis acidic conditions**  described by Lipshutz.<sup>24</sup> Oxidation, as before, yields the title compound.



**In summary, we have described the synthesis of L-phosphoserine analog 5, by two independent routes. Both emanate from readily available building blocks out of the chiral pool. In each case, the critical PCF2-C bond is fashioned by the direct displacement of a primary triflate with 6.10 The synthesis of 5 from (R) isopropylideneglycerol is clearly the more efficient procedure, at this time. Studies on the effectiveness of 5 as a hydmlytically stable phosphoscrinc mimic will be reported in due course.** 

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- **20.**  For 5: <sup>1</sup>H NMR (360 MHz, CDCl<sub>3</sub>):  $\delta$  1.34-1.38 (t, J = 7 Hz, 6 H), 1.42 (s, 9 H), 2.56-2.75 (m, 2 **H), 4.22-4.30 (app quintet, J = 7 Hz, 4 H), 4.55-4.59 (m, 1 H),** *5.37-5.40* **(m, 1 H); 13C NMR (125 MHz, CDCl3): δ 16.27, 16.30 (2 X OCH<sub>2</sub>CH<sub>3</sub>); 28.3 (OCMe3); 35.3-35.4 (br, CH<sub>2</sub>CF<sub>2</sub>); 48.4 (C<sub>α</sub>);** 65.0-65.1 (br, 2 X OCH<sub>2</sub>CH<sub>3</sub>); 80.5 (OCMe<sub>3</sub>); 155.5 (Me<sub>3</sub>COC=O); 173.0 (CO<sub>2</sub>H); HRMS (FAB,  $3-NOBA/Na_2CO_3$ ) calcd for C<sub>13</sub>H<sub>24</sub>F<sub>2</sub>NO<sub>7</sub>PNa  $(M+Na)^+$  398.1156, obsd 398.1155;  $[\alpha]^{25}D^{-10.8^{\circ}}$  $(c 2.20, \text{MeOH})$ . 5 obtained from (R)-isopropylideneglycerol is  $\geq$  96% ee as judged by <sup>1</sup>H NMR [ A **singIe Mosher ester is seen (500 MHz,** *CDCl3)* **following the sequence: (a) CH2N2; (b) HCI, EtGAc; (c) (S)-a-methoxy-a-(trifluoromethyl)phenylacetyl chloride].**
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